

EXPERIMENTAL STUDIES OF EFFICIENCY OF ELECTROMIGRATION TO EXTRACTION OF CHLORIDES

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ABSTRAKT

Příspěvek se zabývá účinností elektromigrace chloridů, která by mohla být použita jako sanační metody pro železobetonových konstrukcí. Experimentální studie testů zrychleného transportu chloridů byla provedena na vzorcích betonu bez chloridů a vzorcích s chloridy přimíchanými do betonu během betonáže. Ve studii byly použity dvě směsi betonu z portlandského cementu charakterizované normální a nízkou pevností v tlaku. K pronikání chloridů do vzorku bez chloridů a pro extrakci chloridů ze vzorku s chloridy byla využita elektromigrace. V práci byla sledována účinnost procesu extrakce chloridů pro sanaci železobetonu z hlediska snížení koncentrace chloridů v různých typech betonu a povrchové koncentraci. Po elektrické extrakci bylo zjištěno snížení počáteční koncentrace chloridu o 15-20 % po 24 hodinách. Snížení povrchových koncentrací bylo v rozmezí 40-50 %. Ukázalo se, že proces extrakce je proveditelný a účinný pro oba typy betonu.

KLÍČOVÁ SLOVA

Beton • Elektromigrace • Chloridy • Extrakce

ABSTRACT

This contribution deals with the efficiency of electromigration of chlorides used as a repair method for reinforced concrete structures. Experimental studies of accelerated chloride transport tests were performed on samples of concrete without chlorides and with admixed sodium chloride during concreting. Two concrete types from Portland cement characterized with normal and low compressive strengths were studied. The electromigration was applied to penetrate chlorides into the chloride-free sample and for extraction of chlorides from the sample. The effectiveness of the chloride extraction process for rehabilitation of reinforced concrete in terms of lowering the chloride concentration in different concrete types and surface concentration was observed. Electrical extraction was found to be effective for lowering of initial chloride concentration by 15-20% after 24 hours. The decrease in surface concentrations was found in the range of 40-50%. The extraction process was found to be feasible and effective for both concrete types.

KEYWORDS

Concrete • Electromigration • Chlorides • Extraction

1. INTRODUCTION

Reinforced concrete structure is exposed to a variety of environmental processes during its lifetime that can cause degradation. The penetration of chlorides into the concrete from deicing salts or in sea splash areas are the most common deterioration processes. Chloride ions diffuse through concrete towards steel reinforcement and cause its corrosion. After reaching a critical value of chloride concentration on the rebar surface the corrosion of the steel starts [1-3]. The corrosion is accompanied by an expansion of the oxidic products causing high pressures, concrete cracking and finally spalling of a cover layer. Therefore, the resistance of concrete to penetration of chlorides is one of the very important properties for steel-reinforced concretes. Since the natural diffusion of ions [4,5] is a slow process taking years further complicated by concrete aging and chloride binding [6] testing of concrete chloride permeability is often done in an accelerated manner using an electric field [7-9]. Both types of tests need to be properly interpreted and diffusion characteristics of concrete can be derived from the tests [9,10].

Electrochemical extraction [7,8] is one of the solutions that can be used to repair reinforced concrete structure already attacked by chlorides. Besides chlorides, other particles with the charge, such as corrosion inhibitors or nanoparticles, can also be transported via electric field [8,11-13]. The convective flux of ions caused by the electromotive force is much larger than the diffusive flux and leads to radical shortening of the time needed for extraction of chlorides from the structure. In the reality, the electric field is realized between the steel reinforcement and the outer surface electrode (usually stainless steel mesh) placed on the concrete surface. In the lab scale, the tests are performed on concrete samples placed in between the mesh electrodes submerged in a suitable electrolyte. A standard test involves rapid chloride penetration [14] which categorizes samples into levels of vulnerability to the chloride attack. More precise measurements with chloride profile assessment will be presented in this paper.

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2. MATERIALS

Two types of concrete mixtures were prepared for this study with low compressive strength (labeled as L) and with normal strength (labeled as N). Each type was prepared without chlorides and with 1% of cement weight NaCl addition to the mixture prepared from ordinary Portland cement CEM I-42,5R, sand and natural crushed aggregate. The mixtures composition is shown in the Table 1. The concrete was mixed in a 50 l laboratory mixer for 10 minutes. The samples were

cylindrical with height of 200 mm and diameter of 100 mm. After casting they were vibrated for approximately 30 seconds. As prevention before water evaporation the specimens were covered with a foil. After 2 days after casting they were unoulded and stored in water. The resulting 28-days compressive strengths and density of the L samples were 24.3 MPa and 2216 kg/m³ and 50.5 MPa and 2340 kg/m³ for N samples, respectively. For chloride penetration tests the specimens were cut to slices with the thickness of 50 mm.

Table 1: Concrete mixtures composition, mass per 1 m³.

Type of concrete	CEM I 42,5R [kg]	Sand 0/4 [kg]	Aggregate 4/8 [kg]	Aggregate 8/16 [kg]	Water [kg]	w/c	NaCl [kg]
N	436.4	872.7	290.9	581.8	186	0.43	0
N-1%	436.4	872.7	290.9	581.8	186	0.43	4.36
L	261.8	1150.0	290.9	581.8	210.2	0.8	0
L-1%	261.8	1150.0	290.9	581.8	210.2	0.8	2.62

3. METHODS

In order to test the effectiveness of the electromigration process two types of accelerated tests were performed in an migration chamber. First, the chloride penetration test was made on the concrete samples without chlorides (N and L series) and second, the accelerated chloride extraction test was performed on the samples with 1 % NaCl per cement addition (N-1% and L-1% series). The migration chamber composed of two containers with electrolyte solutions. In case of chloride

penetration tests, a 3% NaCl solution was used in the compartment with positive electrode and 0.3% NaOH solution in the compartment with negative electrode. In case of chloride extraction, both compartments were filed with 0.3 % NaOH solution and chlorides driven into one of them. A DC power source with constant voltage of 20 V was connected to stainless steel mesh electrodes submerged in the electrolytes. The analyzed samples were inserted between the electrodes and the sample sealed (Fig. 1). The accelerated chloride penetration/extraction run for 24 hours. The scheme of tests is shown in Fig. 2.

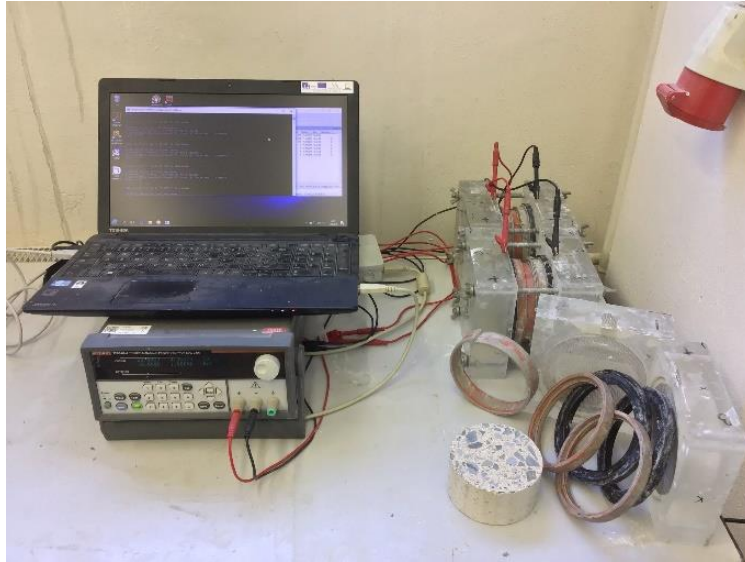


Fig. 1: An electromigration setup used for accelerated chloride penetration/extraction tests.

Before the tests of samples with 1 % chloride addition, the original total chloride concentration was analyzed in 20 mm depth steps by drilling of the powder. Subsequently, the drilled holes were sealed in the samples and samples analyzed further. After the tests, the concentration was analyzed from powder

collected in 5 mm depth steps. Always, the powder was mixed with extraction liquid based on potassium dichromate and acetic acid. After at least 24 hours, the solution was analyzed with Cl ion selective electrode and the chloride profile constructed.

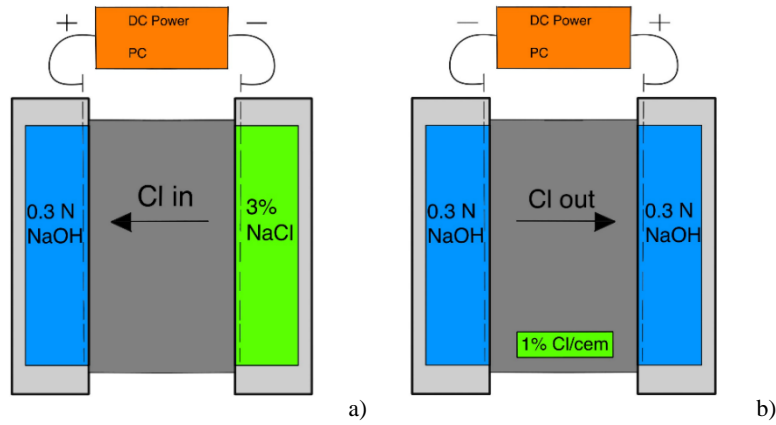


Fig. 2: Scheme of accelerated chloride a) penetration test and b) extraction test.

4. RESULTS AND DISCUSSION

4.1. Penetration of chlorides

The results of individual accelerated chloride penetration tests were used for calculation of average chloride concentration profile from 3-4 samples as shown in Fig. 3 for L samples and

in Fig. 4 for N samples. The profile exhibits a gradual reduction of chlorides from surface of samples, which was exposed to the 3 % NaCl solution. The shape of the profile is different in comparison with traditional diffusion tests. Here, the chlorides are driven by an electromotive force which is dominant and the profile is a "convective" one with the "middle bulge" caused by the convective force.

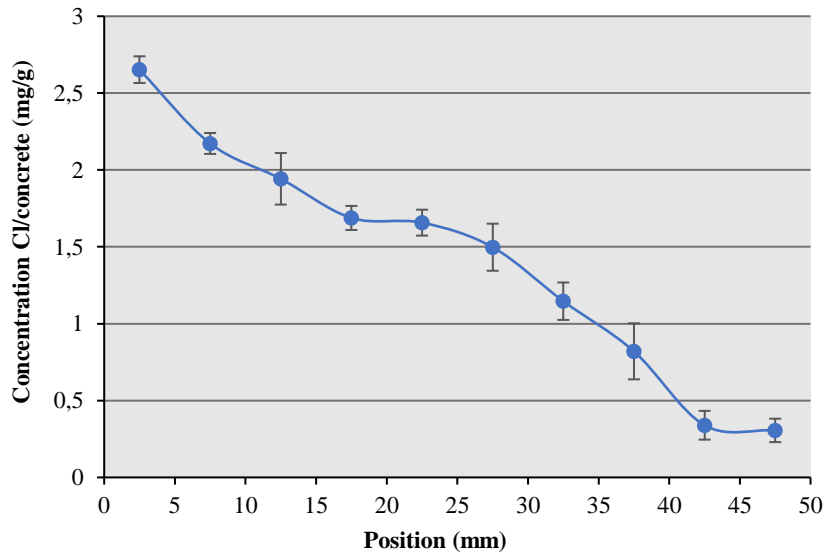


Fig. 3: Concentration profile in L samples.

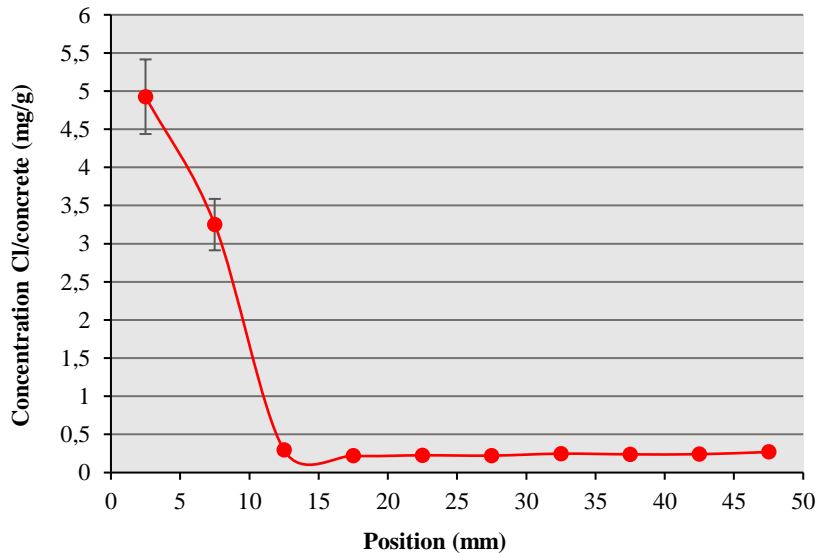


Fig. 4: Concentration profile in N samples.

It can be seen in the figures that the surface concentration after 24 hours exposure is higher by 50% for N samples while the extent of the penetration, i.e. the penetration depth, is larger for L samples. Substantial chloride concentration appears within 40 mm of the L sample while only to 10 mm in the case of N sample. The difference in the surface concentration between L and N samples is related to the amount of cement in the mixture and the ability of the concrete to bind chlorides. The binding is higher for N samples which has twice as cement compared to L samples. The density of the N samples is however higher compared to L samples. Thus the permeability of the N samples is lower and the penetration depth is substantially lower compared to L samples.

4.2. Extraction of chlorides

Original and after-the-extraction chloride concentration profiles are shown in Figs. 5 and 6 for L and N samples,

respectively. Chlorides are extracted from the sample to the right compartment. The extraction causes decrease in concentration for both sample types. Most significantly, the concentration drops down in the surface region (left part of the profile in Figs. 5 and 6). The average decrease of chloride concentration was about 15% on L samples and 20% on N samples. The decrease in the surface region was more the 50 % for L samples and about 40% for N samples.

The profile is in agreement with the convective transport [10] which leads the chloride ions from left of the samples to the right compartment of the migration chamber. Note again, that the transport lasted for 24 hours with voltage of 20 V. Further acceleration can be reached by using higher voltages or using larger times. Increasing voltage is, however, impractical due to safety reasons and due to Joule effect causing heating of the specimen. Increasing of the exposure time is, on the other hand, possible and higher efficiency leading to further lowering of the concentration would take place.

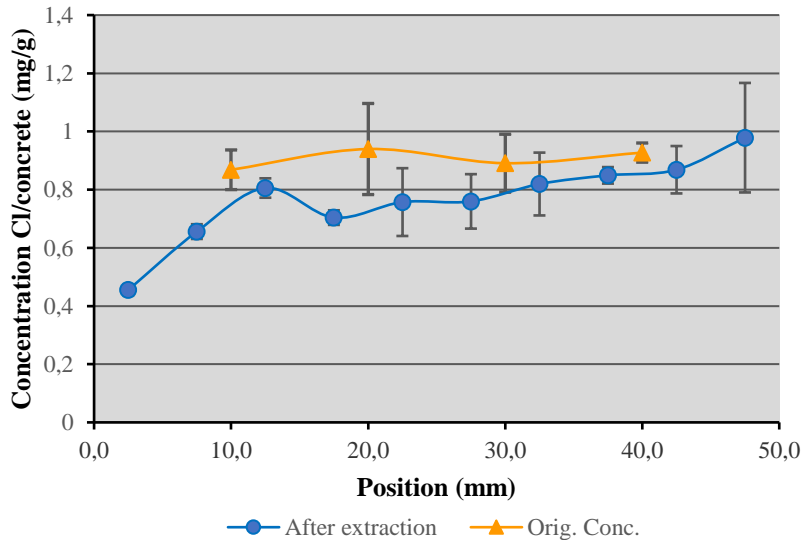


Fig. 5: Concentration profile of L-1% samples before and after extraction.

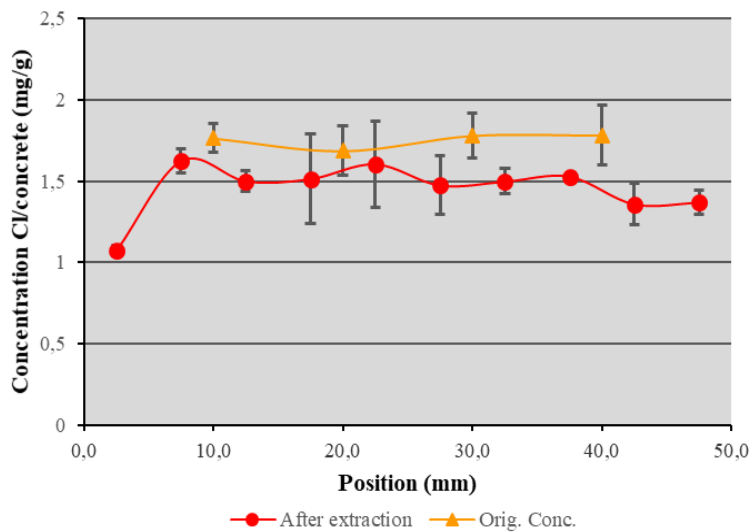


Fig. 6: Concentration profile of N-1% samples before and after extraction.

5. CONCLUSIONS

The presented paper provides results of accelerated chloride penetration and extraction tests performed on two typical concrete mixtures with different compositions with normal and low compressive strengths. The principle of electromigration was applied to penetrate chlorides into the chloride-free sample and to extract chlorides from the sample with some initial Cl concentration. The chloride profiles with concentrations that can be developed after long diffusion tests were gained here in 24 hours while applying safe voltage of 20 V. The chloride concentrations were assessed from concrete core drills using an extraction liquid and ion selective electrode. Typical convective profiles with steep concentration change and different surface concentrations were obtained. The different permeability of the samples caused differences in both surface concentrations as well as in penetration depths. The surface concentration was twice as high in N samples compared to L samples. The penetration depth was three to four times larger in L samples compared to N samples. The first phenomenon can be attributed to a larger binding capacity of N samples given by the twice amount of cement in comparison to L samples. Larger penetration depth in L samples is related to their increased permeability given by their lower density and higher porosity.

Electrical extraction was found to be an effective tool for lowering of initial chloride concentration. 15-20% average concentration decrease was reached after 24 hours. The decrease in surface concentrations was even higher (40-50%) which corresponds to the convection of ions out of the samples. The extraction process was found to be feasible and effective for both concrete types.

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